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Cambridge Assessment International Education

Cambridge O Level

CHEMISTRY **5070/32**

Paper 3 Practical Test **October/November 2022**

CONFIDENTIAL INSTRUCTIONS



This document gives details of how to prepare for and administer the practical exam.

The information in this document and the identity of any materials supplied by Cambridge International are confidential and must NOT reach candidates either directly or indirectly.

The supervisor must complete the report at the end of this document and return it with the scripts.

INSTRUCTIONS

- If you have any queries regarding these confidential instructions, contact Cambridge International stating the centre number, the syllabus and component number and the nature of the query.
  email info@cambridgeinternational.org
  phone +44 1223 553554

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General information about practical exams

Centres must follow the guidance on science practical exams given in the *Cambridge Handbook*.

Safety

Supervisors must follow national and local regulations relating to safety and first aid.

Only those procedures described in the question paper should be attempted.

Supervisors must inform candidates that materials and apparatus used in the exam should be treated with caution. Suitable eye protection should be used where necessary.

The following hazard codes are used in these confidential instructions, where relevant:

C	corrosive	**MH**	moderate hazard
HH	health hazard	**T**	acutely toxic
F	flammable	**O**	oxidising
N	hazardous to the aquatic environment		

Hazard data sheets relating to substances used in this exam should be available from your chemical supplier.

Before the exam

- The packets containing the question papers must **not** be opened before the exam.
- It is assumed that standard school laboratory facilities, as indicated in the *Guide to Planning Practical Science*, will be available.
- Spare materials and apparatus for the tasks set must be available for candidates, if required.

During the exam

- It must be made clear to candidates at the start of the exam that they may request spare materials and apparatus for the tasks set.
- Where specified, the supervisor **must** perform the experiments and record the results as instructed. This must be done **out of sight** of the candidates, using the same materials and apparatus as the candidates.
- Any assistance provided to candidates must be recorded in the supervisor's report.
- If any materials or apparatus need to be replaced, for example, in the event of breakage or loss, this must be recorded in the supervisor's report.

After the exam

- The supervisor must complete a report for each practical session held and each laboratory used.
- Each packet of scripts returned to Cambridge International must contain the following items:
  - the scripts of the candidates specified on the bar code label provided
  - the supervisor's results relevant to these candidates
  - the supervisor's reports relevant to these candidates
  - seating plans for each practical session, referring to each candidate by candidate number
  - the attendance register.

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Specific information for this practical exam

During the exam, the supervisor (**not** the invigilator) must do the experiments in Questions 1 and 2 and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

If chemicals are prepared in more than one batch, clearly labelled supervisor's results must be provided for each batch. The candidates using each batch must be listed on the supervisor's report.

Apparatus

The apparatus listed must be provided to each candidate.

1 × 50 cm^3 burette
1 × 25 cm^3 pipette
1 × pipette filler
1 × stand
1 × burette clamp
1 × funnel for filling burette
1 × white tile
1 × flask or other suitable vessel for titration
a supply of test-tubes
1 × test-tube rack
1 × test-tube holder
1 × stirring rod
2 × boiling tubes
1 × Bunsen burner
1 × wash bottle containing distilled water
1 × heat-proof mat
a supply of teat/dropping pipettes
1 × beaker (for washing teat/dropping pipettes)
paper towels
red and blue litmus papers or universal indicator paper
wooden splints
apparatus normally used in the centre to test for carbon dioxide with limewater

Candidates are expected to rinse and reuse test-tubes and boiling tubes where necessary. Additional tubes should be available.

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Materials

The materials listed in the table must be provided to each candidate. An excess of at least 10% of each material must be prepared to cover accidental loss.

Warning: small amounts of SO_2 **[C] [T]**, which can cause respiratory distress in some people, may be produced. **The laboratory must be well ventilated.**

label		per candidate	identity	notes
	P	150 cm^3	0.10 mol / dm^3 hydrochloric acid	Dilute 8.5 cm^3 of concentrated (37%; approximately 12 mol / dm^3) hydrochloric acid **[C][MH]** in distilled water and make up to 1 dm^3.
	Q	150 cm^3	0.10 mol / dm^3 sodium hydroxide	Dissolve 4.0 g of NaOH **[C]** in distilled water and make up to 1 dm^3.
[F] [HH] [MH] [T] [C] [N]	methyl orange indicator	5 cm^3	methyl orange indicator	See preparation instructions on page 30 of the 2022 syllabus.
Supervisors are asked to carry out a standard acid/base titration between solutions **P** and **Q** to ensure that the concentrations of the two solutions fall within the given range. It is **essential** that 25.0 cm^3 of **Q** reacts with between 24.0 cm^3 and 26.0 cm^3 of **P**				
	R	15 cm^3	0.2 mol / dm^3 iron(II) sulfate in 0.05 mol / dm^3 sulfuric acid	This solution should be freshly prepared. Dissolve 55.6 g of hydrated iron(II) sulfate, $FeSO_4 \cdot 7H_2O$ **[MH]** in 500 cm^3 of distilled water and 50 cm^3 1.0 mol / dm^3 sulfuric acid **[MH]** and dilute the solution with distilled water to 1 dm^3.
[MH] [N]	**S**	15 cm^3	0.1 mol / dm^3 zinc chloride	Dissolve 13.6 g of anhydrous zinc chloride, $ZnCl_2$ **[C] [MH] [N]** in distilled water and make up to 1 dm^3. Hydrogen chloride gas **[C] [T]** is produced when anhydrous zinc chloride is added to water. Use a fume cupboard.
	aqueous hydrogen peroxide	15 cm^3	'10 vol' hydrogen peroxide	Add 100 cm^3 100 volume H_2O_2 **[C]** to distilled water and make up to 1 dm^3.

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label		per candidate	identity	notes
[C]	dilute nitric acid	10 cm^3	1.0 mol / dm^3 HNO_3	See preparation instructions on page 30 of the 2022 syllabus. If necessary, each of these reagents can be provided as a communal supply for groups of up to 6 candidates. Invigilators must be alert to the risk of contamination and the opportunity for malpractice when using a communal supply.
[MH] [N]	aqueous ammonia	30 cm^3	1.0 mol / dm^3 NH_3	
[C]	aqueous sodium hydroxide	30 cm^3	1.0 mol / dm^3 NaOH	
	aqueous barium chloride or aqueous barium nitrate	10 cm^3	0.1 mol / dm^3 $BaCl_2$ or 0.1 mol / dm^3 $Ba(NO_3)_2$	
	aqueous silver nitrate	10 cm^3	0.05 mol / dm^3 $AgNO_3$	
[MH]	limewater	10 cm^3	saturated aqueous calcium hydroxide, $Ca(OH)_2$	

- All solutions must be thoroughly mixed.
- Materials must be labelled only as specified in the 'label' column. The identities of chemicals labelled with letter codes, e.g. **P**, may be different from their descriptions in the question paper. Candidates must use the descriptions given in the question paper.
- If you are unable to source any of these chemicals, you must contact Cambridge International as far as possible in advance of the exam for advice.

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Supervisor's report

Syllabus and component number | | | | | / | | |

Centre number | | | | | |

Centre name .....

Time of the practical session .....

Laboratory name/number .....

Give details of any difficulties experienced by the centre or by candidates (include the relevant candidate names and candidate numbers).

You must include:
- any difficulties experienced by the centre in the preparation of materials
- any difficulties experienced by candidates, e.g. due to faulty materials or apparatus
- any specific assistance given to candidates.

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If chemicals have been prepared in more than one batch, list the candidates using each batch.

Declaration

1 Each packet that I am returning to Cambridge International contains all of the following items:
- the scripts of the candidates specified on the bar code label provided
- the supervisor's results relevant to these candidates
- the supervisor's reports relevant to these candidates
- seating plans for each practical session, referring to each candidate by candidate number
- the attendance register.

2 Where the practical exam has taken place in more than one practical session, I have clearly labelled the supervisor's results, supervisor's reports and seating plans with the time and laboratory name/number for each practical session.

3 I have included details of difficulties relating to each practical session experienced by the centre or by candidates.

4 I have reported any other adverse circumstances affecting candidates, e.g. illness, bereavement or temporary injury, directly to Cambridge International on a *special consideration form*.

Signed ..... (supervisor)

Name (in block capitals) .....

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