

## **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

#### **COMBINED SCIENCE**

Paper 1 Multiple Choice (Core)

0653/12 October/November 2018 45 minutes

Additional Materials: Multiple Choice Answer Sheet Soft clean eraser Soft pencil (type B or HB is recommended)

### READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, glue or correction fluid. Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you. DO **NOT** WRITE IN ANY BARCODES.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A**, **B**, **C** and **D**.

Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

#### Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer. Any rough working should be done in this booklet. A copy of the Periodic Table is printed on page 20. Electronic calculators may be used.

This document consists of 18 printed pages and 2 blank pages.



**1** Movement is a characteristic of all living organisms.

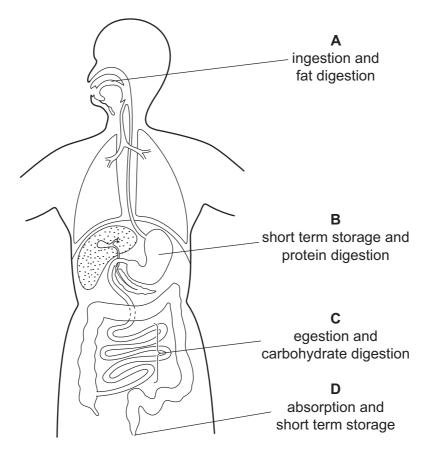
Which two other characteristics of living organisms provide the energy for movement?

- A excretion and nutrition
- **B** growth and sensitivity
- **C** nutrition and respiration
- **D** respiration and growth
- 2 Which process depends on diffusion?
  - **A** circulation
  - **B** digestion
  - **C** gaseous exchange
  - D phagocytosis
- 3 Enzymes are used in digestion to break down larger molecules into smaller molecules.

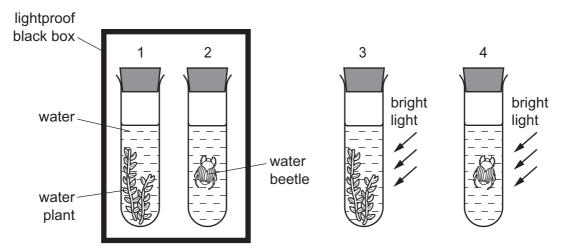
Which row matches the large molecules with the small molecules they are broken down into?

	large molecules	small molecules
Α	fat	glycerol and fatty acids
в	glycogen	glycerol and amino acids
С	protein	simple sugars
D	starch	amino acids

4 The diagram shows the human alimentary canal. The labels state the functions of four of its parts. Which label is correct?



**5** The diagram shows apparatus set up to investigate photosynthesis.



In which test-tubes will the carbon dioxide concentration in the water decrease?

	test-tube			
1 2 3 4				4
Α	no	no	yes	no
В	no	no	yes	yes
С	yes	no	no	no
D	yes	yes	no	no

6 The table shows components of blood and their functions.

#### Which row is correct?

	blood component	function of component
Α	plasma	antibody formation
в	platelets	transport of carbon dioxide
С	red blood cells	blood clotting
D	white blood cells	phagocytosis

- 7 Which word equation represents aerobic respiration?
  - A carbon dioxide + water  $\rightarrow$  glucose
  - **B** carbon dioxide + water  $\rightarrow$  glucose + oxygen
  - $\textbf{C} \quad \text{glucose} \ \textbf{+} \ \text{oxygen} \ \rightarrow \ \text{carbon dioxide}$
  - **D** glucose + oxygen  $\rightarrow$  carbon dioxide + water

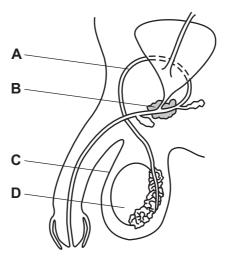
**8** What happens to the blood glucose concentration and pulse rate when adrenaline is released into a person's bloodstream?

	blood glucose concentration	pulse rate
Α	decreases	increases
в	decreases	decreases
С	increases	increases
D	increases	decreases

**9** A plant produces flowers with stigmas and ovaries, but no anthers.

What effect will this have on the plant?

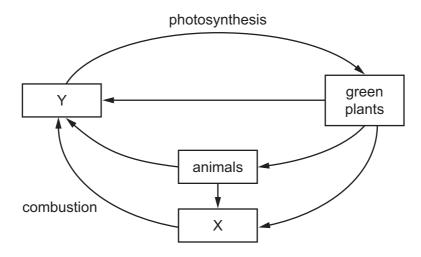
- **A** It cannot be insect-pollinated.
- B It cannot produce seeds.
- **C** It will only be able to reproduce asexually.
- **D** It will not be able to produce pollen grains.
- **10** The diagram shows the male reproductive system.



Which row names the labelled structure and states its function?

	name of structure	function of structure
Α	sperm duct	sperm pass directly into the uterus from this tube
В	testis	produces the sperm
С	scrotum	holds the testis
D	prostate gland	stores the sperm

- **11** What is a herbivore?
  - A an organism that gets its energy by eating other animals
  - **B** an organism that gets its energy by eating plants
  - **C** an organism that makes its own organic nutrients
  - **D** the first organism in a food chain
- **12** The diagram shows part of the carbon cycle.



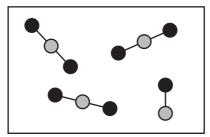
What are X and Y?

	Х	Y
Α	carbon dioxide	oxygen
в	fossil fuel	carbon dioxide
С	fossil fuel	oxygen
D	oxygen	carbon dioxide

13 Which are possible harmful effects of deforestation?

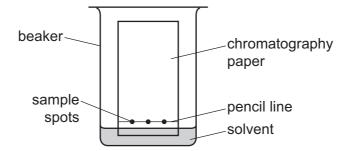
	global warming	species extinction
Α	$\checkmark$	1
в	$\checkmark$	×
С	×	1
D	x	x

14 The diagram represents a mixture of carbon dioxide, CO<sub>2</sub>, and carbon monoxide, CO.



Which statement is correct?

- **A** The mixture contains 4 elements.
- **B** The mixture contains 4 molecules.
- **C** The mixture contains 11 elements.
- **D** The mixture contains 11 molecules.
- **15** The apparatus used for chromatography is shown.



Which statement about the method used for chromatography is not correct?

- **A** The beaker is swirled to help the solvent to rise.
- **B** The chromatography paper is placed in the beaker after the solvent has been added.
- **C** The chromatography paper is removed before the solvent reaches the top of the paper.
- **D** The sample spots are placed on the pencil line above the level of the solvent.
- 16 The atomic number of element X is 11.

The mass number of element X is 23.

Which statement about an atom of X is correct?

- A It contains 12 nucleons in the nucleus and 11 orbiting electrons.
- **B** It contains 12 nucleons in the nucleus and 11 orbiting protons.
- C It contains 23 nucleons in the nucleus and 11 orbiting electrons.
- **D** It contains 23 nucleons in the nucleus and 11 orbiting protons.

17 Which formula does not represent an acid?

**A**  $H_2SO_4$  **B** HCl **C**  $HNO_3$  **D**  $NH_3$ 

**18** The breakdown of molten lead bromide by .....1..... forms the elements lead and bromine.

Lead is formed at the .....2.....

Which words complete gaps 1 and 2?

	1	2
Α	electrolysis	anode
в	electrolysis	cathode
С	reduction	anode
D	reduction	cathode

**19** Sodium chloride dissolves in water in an endothermic process.

When calcium chloride dissolves in water, the temperature increases.

Which statement is correct?

- **A** The process of dissolving calcium chloride is neither exothermic nor endothermic.
- **B** The temperature increases when sodium chloride dissolves.
- **C** The temperature remains constant when sodium chloride dissolves.
- **D** When calcium chloride dissolves in water the process is exothermic.
- 20 Hydrogen peroxide decomposes to form oxygen and water.

A catalyst is added to the hydrogen peroxide.

Which row describes the change in the rate of reaction and the mass of catalyst left at the end of the reaction?

	rate of reaction	mass of catalyst left at end of reaction
Α	decrease	less
в	decrease	no change
С	increase	less
D	increase	no change

**21** Iron oxide reacts with carbon monoxide.

The word equation for the reaction is:

iron oxide + carbon monoxide  $\rightarrow$  iron + carbon dioxide

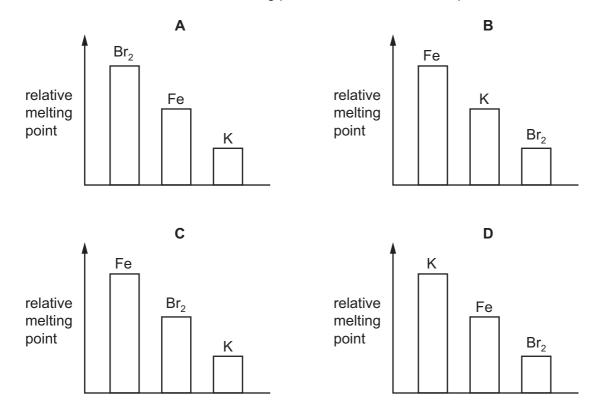
Which statement is not correct?

- A Carbon is neither oxidised nor reduced.
- B Carbon is oxidised.
- C Iron is reduced.
- **D** This is a redox reaction.
- 22 Compound X reacts with compound Y to form sodium sulfate, water and carbon dioxide.

What are X and Y?

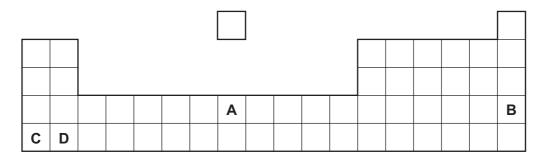
	Х	Y
Α	sodium carbonate	hydrochloric acid
В	sodium carbonate	sulfuric acid
С	sodium hydroxide	hydrochloric acid
D	sodium hydroxide	sulfuric acid

23 Which bar chart shows the relative melting points of bromine, iron and potassium?



**24** The positions of four elements are shown in the outline of the Periodic Table.

Which element has a high melting point and forms coloured compounds?



25 Metal X reacts rapidly with steam but only very slowly with cold water.

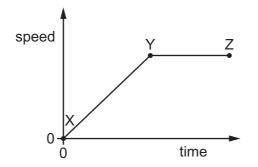
What is X?

- A copper
- B iron
- C magnesium
- D sodium
- 26 What is a chemical test for water?
  - **A** Add blue cobalt chloride paper.
  - **B** Add blue copper sulfate crystals.
  - C Measure the density.
  - **D** Measure the melting point.
- **27** Gasoline is a hydrocarbon fuel obtained from petroleum.

Which statement is correct?

- A Gasoline burns to form carbon dioxide and water.
- **B** Gasoline contains the elements carbon, hydrogen and oxygen.
- **C** Gasoline is used as a fuel in diesel engines.
- **D** The combustion of gasoline is an endothermic reaction.

**28** A car is travelling along a straight road. The diagram is a speed-time graph for part of its journey.



What is happening to the car between X and Y and what is happening between Y and Z?

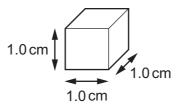
	between X and Y	between Y and Z
Α	changing speed	constant speed
В	changing speed	not moving
С	constant speed	constant speed
D	constant speed	not moving

**29** An object of known mass on Earth is taken to another planet.

Which row gives correct information about the mass of the object on the other planet?

	unit of mass	mass of object
Α	kilogram	different
В	kilogram	the same
С	newton	different
D	newton	the same

**30** A cube of aluminium has sides of length 1.0 cm.



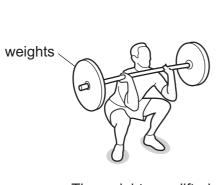
Compared with this cube, which statement about a cube of aluminium with sides of 2.0 cm is correct?

- A It has the same density.
- **B** It has the same mass.
- **C** It has twice the density.
- **D** It has twice the mass.

**31** Weightlifting involves a number of different stages.

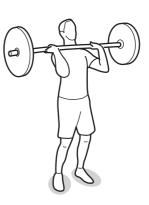
In which stage is **no** work being done on the weights?

Α



The weights are lifted up off the floor.





The weights are lifted as the man stands up.



The weights are lifted above the head.



The weights are held stationary above the head.

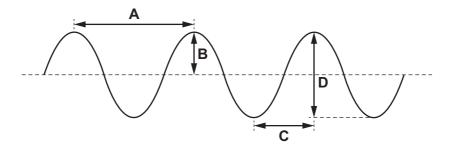
**32** A liquid evaporates when molecules leave its surface.

Which molecules leave the surface, and what happens to the temperature of the remaining liquid?

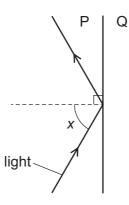
- **A** The more energetic molecules leave and the temperature falls.
- **B** The more energetic molecules leave and the temperature rises.
- **C** The less energetic molecules leave and the temperature falls.
- **D** The less energetic molecules leave and the temperature rises.

**33** The diagram represents a wave.

Which labelled distance gives the amplitude of the wave?



**34** The diagram shows light travelling in substance P. The light strikes substance Q at an angle of incidence *x*. The light is totally internally reflected.



One of the substances is air and the other substance is glass.

Which substance is air, and how does angle x compare with the critical angle?

	substance that is air	angle x
Α	Р	greater than critical angle
в	Р	less than critical angle
С	Q	greater than critical angle
D	Q	less than critical angle

**35** A student is watching television. He uses a remote controller to change the programme.

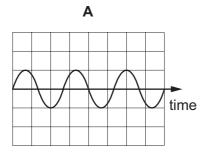
The remote controller uses electromagnetic waves. Electromagnetic waves are also used to transmit the television signals from a satellite.

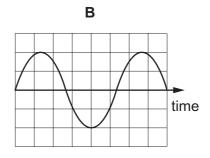
Which row shows the type of wave used for each purpose?

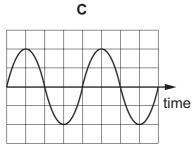
	remote controller	satellite
Α	infra-red	microwaves
в	infra-red	radio waves
С	ultraviolet	microwaves
D	ultraviolet	radio waves

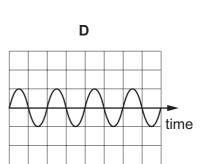
**36** The diagrams represent four different sound waves. The scales are the same in all the diagrams.

Which sound has the lowest pitch?



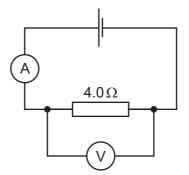






- 37 Which group contains only electrical insulators?
  - A air, lead, rubber
  - B copper, lead, steel
  - **C** plastic, rubber, wood
  - D plastic, steel, wood

**38** The circuit shown includes a  $4.0 \Omega$  resistor.



Which pair of meter readings in the table is possible?

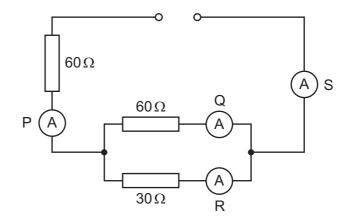
	voltmeter reading / V	ammeter reading / A
Α	0.50	2.0
В	1.0	4.0
С	2.0	0.5
D	2.5	1.5

**39** It is hazardous to use an electric hairdryer near a hot bath.

Why is this?

- **A** The cable to the hairdryer might overheat.
- **B** The motor in the hairdryer might become rusty and not work.
- **C** The water might enter the hairdryer so the user could receive an electric shock.
- **D** The warmth of the atmosphere might damage the insulation in the hairdryer.

**40** The circuit contains four ammeters P, Q, R and S.



Which statement about the readings on the ammeters is correct?

- **A** The reading on S is less than the reading on P.
- **B** The reading on Q is greater than the reading on S.
- **C** The reading on R is greater than the reading on S.
- **D** The reading on Q is less than the reading on P.

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 $\parallel$  $\equiv$  $\leq$ >  $^{6}$  C carbon carbon  $^{120}$  L C  $^{228}$  Silicon  $^{$  $\geq$  $\begin{array}{c} 5 \\ B \\ boron \\ 111 \\ 113 \\ 113 \\ 113 \\ 127 \\ 227 \\ 227 \\ 227 \\ 227 \\ 227 \\ 227 \\ 227 \\ 138 \\ 115 \\ 1$  $\equiv$ Group 27 CO 59 59 77 103 109 109 109 109 109 109 109 109 109 - T <sup>1</sup> 25 26 55 55 57 75 75 75 75 75 107 107 107 107 24 Cr 52 52 96 96 96 96 106 1106 1106 106 Sg atomic symbol name relative atomic mass atomic number Key 21 SC candium 45 45 39 39 89 89 39–103 actinoids A Beryllium beryllium beryllium beryllium beryllium beryllium beryllium agnesium agnesium agnesium beryllium berylli =  $[] \begin{array}{c} \textbf{I} \quad \textbf{I} \quad$ 

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The volume of one mole of any gas is 24 dm<sup>3</sup> at room temperature and pressure (r.t.p.).

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**The Periodic Table of Elements**