

## **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/22

Paper 2 Core Theory

October/November 2016

MARK SCHEME
Maximum Mark: 80

## **Published**

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[1]

[1]

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1

(a) radio (waves) in RH box; [1] [1] (b) (i) cell/battery; (ii) chemical (energy); [1] (c) (i) kinetic; [2] sound; (ii) (higher pitch) A and (larger amplitude) A; [1] (d) (i) any one from: damp conditions/water; damaged insulation (in unit); current too high/could overheat/cause a fire; [1] (ii) fuse; [1] (e) (i) lamp at least two diverging rays from a point on lamp to lens, then emerging from

lens parallel (as shown, arrows not required);

(ii) focal length;

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2 (a)  $C_2H_5OH/C_2H_6O$  any order/ $CH_3CH_2OH$ ;

[1]

(b) (ethanol) + oxygen → carbon dioxide + water LHS; RHS;

[2]

(c)

	test result		
carbon dioxide	limewater ;	(turns) cloudy;	
oxygen	glowing splint ;	relights;	

[4]

(d) increases; [1]

(e) fractional distillation; [1]

3 (a) A cell wall;

**B** chloroplast;

C vacuole;

(b) (i) cuticle correctly labelled on diagram; [1]

(ii) cell drawn right way up in palisade layer; [1]

(c) sugar/glucose + oxygen; [1]

(d) carbon dioxide - any two from:

by diffusion;

through the stomata/intercellular spaces;

from the air;

water - any two from:

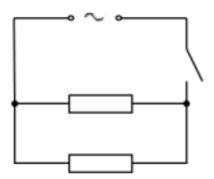
through the xylem;

from the roots/by the transpiration stream;

from the soil; [4]

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## 4 (a)



resistor **and** switch symbols; resistors in parallel; supply, switch, in series;

[3]

(b) (i) conduction;

[1]

(ii) density = mass/volume or d = m/V or V = m/d or 128/8; = 16 (cm<sup>3</sup>);

[2]

(iii) (thickness = volume/area = 16/160) = 0.10 (cm)

[1]

(c)

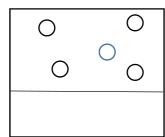


diagram shows example only – look for four similar-sized circles placed randomly <u>apart</u> from each other and from the given circle;

[1]

(d) metals expand on heating; brass expands more than steel; so bends and breaks contact;

[max 2]

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**5** (a) anode;

cathode;

electrolyte; [3]

(b) chlorine;

copper; [2]

(c) (i) copper hydroxide/copper carbonate (/copper sulphide); [1]

(ii) increase temperature/increase concentration/catalyst/decrease particle size; [1]

(d) any two from:

(copper) forms coloured compounds; (copper) has higher melting point/boiling point; copper/copper compounds act as catalyst(s); AVP

[2]

[2]

(e) (bronze is) harder/stronger; [1]

**6** (a) arrow drawn going from plasma into alveolus; [1]

**(b) (i)**  $0.6 \, \text{dm}^3$ 

(ii)  $(0.6 \times 3) = 1.8 \,\mathrm{dm}^3$ 

(c) became faster; became deeper;

(d) any two from:

muscle contraction; protein synthesis;

cell division; growth;

passage of nerve impulses;

maintenance of body temperature ; [2]

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Syllabus Pape
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22

7	(a)	(i)	newton ;		[1]
		(ii)	weight/gravitat	ional force ;	[1]
	(b)	(i)		t (45,15) and (60, 20) +/– half a small square ; nded to at least to (60, 20) ;	[2]
		(ii)	answer in range	e 24 (cm) to 30 (cm);	[1]
	(c)	whe	O(N); en cords are fully anced / <i>owtte</i> ;	stretched, no further movement/change in length/forces	[2]
8	(a)	no	new substance r	made/no chemical reaction occurs ;	[1]
	(b)	any	npound/molecul one from: taining hydrogel y;		[2]
	(c)	(ga	înery gas) soline) s oil)	heating/cooking; AVP car fuel/petrol; AVP lorry fuel/bus fuel/diesel; AVP	[3]
	(d)		C bond shown (1 y correct structur		[2]

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[1]

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(a) (a network of) interconnected food chains; showing energy flow (through part of an ecosystem);
(b) Sun; producers; consumers; water flea; turtle;
(c) (i) (algae) increase less being eaten;

(ii) (large fish) decrease less food;