

**CAMBRIDGE INTERNATIONAL EXAMINATIONS**

Cambridge International General Certificate of Secondary Education

**MARK SCHEME for the May/June 2015 series**

**0653 COMBINED SCIENCE**

**0653/63**

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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1 (a) yeast dead / (enzyme) no longer active / denatured ; [1]

(b)

<b>time / mins</b>	<b>colour in tube A</b>	<b>colour in tube B</b>	<b>colour in tube C</b>
1	blue	blue	blue
2	<b>colourless</b>	<b>blue</b>	<b>blue</b>
3	<b>colourless</b>	<b>blue</b>	<b>blue</b>
4	<b>colourless</b>	<b>blue</b>	<b>blue</b>
5	<b>colourless</b>	<b>colourless</b>	<b>blue</b>
6	<b>colourless</b>	<b>colourless</b>	<b>blue</b>

time / mins ;

**A** correct ;

**B** correct ;

**C** correct ;

ALLOW decolourised IGNORE transparent

[4]

(c) (i) constant volume / concentration ;

[1]

(ii) A changes quicker / changes first / respire faster ;  
(more) glucose / substrate available in A ;

[2]

M2 dependent on times being considered

(d) (colour changes back to) blue ;  
methylene blue oxidised / reacts with oxygen / oxygen introduced ;  
oxygen from air above solution ;

[max2]

**[Total: 10]**

2 (a) make a solution in water ;  
add (aqueous) sodium hydroxide / (aqueous) ammonia ;  
green (gelatinous) ppt / solid ;

[3]

(b) add sodium hydroxide (solution) and heat ;  
damp ;  
(red) litmus turns blue ;

[3]

(c) make a solution in water ;  
add hydrochloric / nitric acid ;  
add barium chloride / nitrate (solution) ;  
white ppt ;

[4]

**[Total: 10]**

Page 3	Mark Scheme	Syllabus	Paper
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- 3 (a) correct symbol for voltmeter ;  
connected in parallel between **X** and **Y** or equivalent ; [2]
- (b) (i) values in table: [2]  
1.81 ; ALLOW range 1.80 – 1.82  
0.70 ;
- (ii) headings: V, A,  $\Omega$  (all three required) ; [1]
- (iii) 3.91, 8.00, 2.59 (allow ecf on third value)  
all values to 2 d.p ;  
all correct values ; [2]
- (c) use of 3.91 and 2.59 ;  
statement matches results (expect NO)  
**AND**  
justification: e.g. values are too different/not close enough, even allowing  
for experimental error/is 1.5 times ; [2]
- (d) the lamps are at different temperatures/lamps have different resistances or currents  
than expected/this could explain why teacher statement not supported ; [1]

**[Total: 10]**

- 4 (a) (i) 61 ; [1]
- (ii) 433 ; [1]
- (iii) 0.0023 ; [1]
- (b) (i) Correct plotting (allow 1 error) ;  
SMOOTH curve ; [2]
- (ii)  $52 \pm 2$  ; [1]
- (iii) Do not know the rate either side of  $52^{\circ}\text{C}$ /need more results  
in range e.g.  $40^{\circ}\text{C}$  to  $60^{\circ}\text{C}$  ; [1]
- (c) repeat experiment with water instead of acid ;  
 $1\text{ cm}^3$  ;  
solution will remain cloudy ; [3]

**[Total: 10]**

Page 4	Mark Scheme	Syllabus	Paper
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- 5 (a) (i) lamp/bulb/ammeter ; [1]
- (ii) correct symbol for cell (or battery) ; [1]
- (iii) (explanation) does not react ;  
(material) e.g. carbon/platinum ; [2]
- (b) (i) gives red-brown ppt ; [1]
- (ii) damp litmus ;  
(red then) bleached ; [2]
- (iii) hydrogen ;  
lit splint ;  
“pops” ; [3]

[Total: 10]

- 6 (a) (i) 21.5 ; 20.5 ; [2]
- (ii) axes correct and labelled ;  
vertical axis NOT starting at zero ;  
points correct (allow 1 error) ; (e.c.f. from part (i)) [3]
- (iii) no, points scattered / no pattern / no straight line ; (e.c.f. from parts (i) and (ii))  
(ignore any line drawn) [1]
- (b) (any **three** of)  
rods should be same length and width ;  
amount of wax should be the same ;  
experiment repeated and average taken ;  
water should be stirred ; [3]
- (c) (answer depends upon (b))  
keep thickness / length (etc.) means only variable is % magnesium ;  
repeating identifies anomalous results ; [1]

[Total: 10]