UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2006 question paper

0620 CHEMISTRY

0620/06

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

The grade thresholds for various grades are published in the report on the examination for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses.

CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the October/November 2006 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



www.dynamicpapers.com

Page 2	Mark Scheme	Syllabus	Paper	
	IGCSE - OCT/NOV 2006	0620	6	

1 (a) Boxes filled in correctly to show tongs(1) watch glass/evaporating basin/dish(1) beaker(1) [3] (b) oxidation/combustion/exothermic/redox(1) [1] (c) > 7(1) [1] 2 (a) brown/orange(1) [1] (b) oxygen used in rusting(1) not air [1] $\frac{25}{150} \times 100 \text{ (1)} = 17\%/16.6 \rightarrow 17\%(1) \text{ 2 for correct answer}$ [2] (d) more rust/quicker to rust/water further up tube/tube fills up(1) [1] 3 table of results all volumes correct (2) 0, 9, 35, 62, 81, 88, 89 [2] -1 for any incorrect (a) graph points (2) S-shaped curve joining all points(1) [3] exothermic/displacement/oxidation/redox(1) (b) [1] (c) slow at start/speeds up/slows down at end max 2 [2] (ii) surface dirty owtte at start/then clean/calcium being used up/warms up max 2 [2]

www.dynamicpapers.com

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0620	6

4

5

6

	table of results	6				
(a)	temperatures	corre	ctly con	npleted(3) differences(1)	
	zinc	24	56	32		
	iron	25	41	16		
	magnesium	23	69	46		
	-1 for each inc	orrec	t			[4]
(b)	(i) magnesi	ium(1)			[1]
	(ii) gas evol	ved ra	apidly/r	eacts(1)	greatest (temperature) difference(1)	[2]
	(iii) hydrogei	n(1)				[1]
(c)	Table of result	s ten	nperatu	ıres corr	rect (6)	[6]
	Time /e	_	zin o		magnasium	
	Time /s 0		zinc 24		magnesium 26	
	10		27		54	
	20		29		62	
	30		33		67	
	40		37		68	
	50 60		40 43		67 65	
			.0			
(d)	Graph. Points	plotte	ed corre	ectly(2) -	- 1 for each incorrect	
	Smooth lines(1) lab	pels(1)			[4]
(e)	temperature a	fter 5	s/ 25-2	6°C ± 0.	5°C(1) indication on grid(1)	[2]
(f)	sketch line for	Mg b	elow p	owder M	1 g(1)	
	sketch line for	iron p	owder	below z	zinc/ any line below top curve(1)	[2]
(g)	prevent heat lo	oss/in	sulatio	n(1)		[1]
(h)	one improvem	ent e	.g. use	a burette	e/pipette to measure solution/ lid(1)	[1]
(a)	(ii) red(1) lit	mus t	urns bl	ue(1) re	eference to smell(1) max 2	[2]
(c)	weak(1) acid(1)				[2]
(d)	ammonia(1)					[1]
(e)	ammonium ch	loride	(1)			[1]
(f)	potassium iodi	ide(1)				[1]
(a)	•		` ,	` '	olid residue(1) max 2	[2]
(b)	solid + organic	solve	ent(1)	add to p	paper(1)	
	chromatograpl	hy(1)	use of	solvent	(1) description of spots(1)	
	max 4 NB use	e of w	ater = ı	max 1 fo	or chromatography	[4]
(c)	(i) apply pa	int, st	art time	er(1) me	ethod of checking dry, note time(1)	
	no painti	ng =	0			[2]
	(ii) correct n	netho	d(1) e	.g. hair d	drier/wind/fan/increase temperature.	
	<u>NOT</u> cat	alyst.				[1]

www.dynamicpapers.com

Page 4	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0620	6

[Total 60]